

Large D limit for excited states of a hydrogen atom

I. FORMULATION OF THE PROBLEM

We are going to prove that as $D \rightarrow \infty$,

$$E \sim V_{\min}, \quad (1)$$

where E is the energy and V_{\min} is the minimum of the function

$$V_{\text{eff}}(r) = \frac{n^2}{2r^2} - \frac{1}{r} \quad (2)$$

at $r > 0$. In equation (2), n is the principal quantum number.

A. Definition of principal quantum number in D dimensions

In three dimensions, the principal quantum number of a given quantum state is usually defined as a sum of “orbital quantum number” and “radial quantum number” plus one, i.e.

$$n = l + n_r + 1, \quad (3)$$

or alternatively just as $1/\sqrt{-2E}$. If we adopt the latter definition of the principal quantum number in D dimensions, i.e.

$$n = \frac{1}{\sqrt{-2E}}, \quad (4)$$

then there is nothing to prove, since obviously the function (2) has a minimum

$$V_{\min} = -\frac{1}{2n^2} = E \quad (5)$$

at the point

$$r_{\min} = n^2. \quad (6)$$

We adopt here the following equivalent definition of the quantum number n ,

$$n = l + n_r + \frac{D - 1}{2}, \quad (7)$$

where n_r is the number of positive roots of the radial wavefunctions, and l is some sort of generalization of orbital quantum number to D dimensions.

To avoid complications related to angular momentum in arbitrary dimensions, we consider here only the case of zero angular momentum ($l = 0$) by solving the problem inside a subspace of radially-symmetric wavefunctions. Without knowledge of the form of exact solution, we find in Section IV that asymptotically at large D , $E/V_{\min} \rightarrow 1$, or more precisely $E/V_{\min} - 1 \sim o(1/n)$. In fact, if we would know the energy exactly, we could see that $E = V_{\min}$, but the strength of current approach is the proof *without* solving the equation exactly.

II. RADIAL SCHRÖDINGER EQUATION

We assume that the function $\psi(r)$ depends only on one variable which is radius $r = (\mathbf{r}^2)^{1/2}$. Derivatives of the wave function are as follows,

$$\begin{aligned} \frac{\partial \psi(r)}{\partial r_i} &= \frac{r_i}{r} \psi'(r), \\ \frac{\partial^2 \psi(r)}{\partial r_i^2} &= \frac{1}{r} \psi'(r) + \frac{r_i^2}{r^2} \left(\psi''(r) - \frac{1}{r} \psi'(r) \right). \end{aligned} \quad (8)$$

Using (8), the kinetic energy is expressed as

$$-\frac{\nabla^2}{2} \psi(r) = -\frac{1}{2} \sum_{i=1}^D \frac{\partial^2 \psi(r)}{\partial r_i^2} = -\frac{1}{2} \left[\psi''(r) + \frac{D-1}{r} \psi'(r) \right]. \quad (9)$$

Thus, the Schrödinger equation is reduced to a differential equation in one variable r ,

$$-\frac{1}{2} \left[\psi''(r) + \frac{D-1}{r} \psi'(r) \right] - \frac{1}{r} \psi(r) = E \psi(r), \quad (10)$$

where we use units where $\hbar = m = 1$.

It is convenient to introduce “radial” function $P(r) = r^{\frac{D-1}{2}} \psi(r)$ and to rewrite equation (10) in an equivalent form without linear derivatives,

$$-\frac{1}{2} \frac{d^2 P}{dr^2} + \left[\frac{(D-1)(D-3)}{8r^2} - \frac{1}{r} - E \right] P(r) = 0. \quad (11)$$

Using equation (7), we express D through n as $D = 2n - 2n_r + 1$ and find that

$$\frac{(D-1)(D-3)}{8r^2} = \frac{(n-n_r)(n-n_r-1)}{2r^2}. \quad (12)$$

III. SCALING TRANSFORMATION

Let us perform scaling transformation $r = n^2(1 + n^{-1/2}x)$. Scaled wave function y is defined as

$$y(x) = P \left(n^2(1 + n^{-1/2}x) \right). \quad (13)$$

Then,

$$\frac{d^2 P}{dr^2} = n^{-3} \frac{d^2 y}{dx^2}, \quad (14)$$

where l.h.s. is calculated at the point $r = n^2(1 + n^{-1/2}x)$.

We rewrite

$$\frac{(n-n_r)(n-n_r-1)}{2r^2} - \frac{1}{r} - E = \frac{(r-n^2)^2}{2n^2 r^2} - \frac{(2n_r+1)n}{2r^2} + \frac{n_r(n_r+1)}{2r^2} - \tilde{E}, \quad (15)$$

where

$$\tilde{E} = E + \frac{1}{2n^2}. \quad (16)$$

Now, we express each term in equation (15) through a new variable x ,

$$\frac{(r - n^2)^2}{2n^2r^2} = n^{-3} \frac{x^2}{2(1 + n^{-1/2}x)^2}, \quad (17)$$

$$-\frac{(2n_r + 1)n}{2r^2} = -n^{-3} \frac{2n_r + 1}{2(1 + n^{-1/2}x)^2}, \quad (18)$$

$$-\frac{n_r(n_r + 1)}{2r^2} = n^{-4} \frac{n_r(n_r + 1)}{2(1 + n^{-1/2}x)^2}. \quad (19)$$

After multiplying equation (11) by n^3 and using (13) - (19), we find

$$\left[-\frac{1}{2} \frac{d^2}{dx^2} + \frac{x^2}{2(1 + n^{-1/2}x)^2} - \frac{n_r + 1/2}{(1 + n^{-1/2}x)^2} + n^{-1} \frac{n_r(n_r + 1)}{2(1 + n^{-1/2}x)^2} - \epsilon \right] y(x) = 0, \quad (20)$$

where

$$\epsilon \equiv n^3 \tilde{E} = n^3 E + \frac{n}{2} = -\frac{n}{2} \left(\frac{E}{V_{\min}} - 1 \right). \quad (21)$$

IV. LARGE- n LIMIT OF SCALED SCHRÖDINGER EQUATION

In the limit of $n \rightarrow \infty$, equation (20) reads

$$\left[-\frac{1}{2} \frac{d^2}{dx^2} + \frac{1}{2} x^2 - (n_r + 1/2) - \epsilon_\infty \right] y_\infty(x) = 0. \quad (22)$$

It is the equation for a harmonic oscillator shifted by a constant $-(n_r + 1/2)$ on the energy scale. It is known that the oscillator's energy eigenvalue equals to $n_r + 1/2$. After adding the extra constant $-(n_r + 1/2)$, we obtain $\epsilon_\infty = 0$.

Thus, we proved that

$$n \left(\frac{E}{V_{\min}} - 1 \right) \rightarrow 0, \quad (23)$$

i.e. $E/V_{\min} - 1 \sim o(1/n)$.

Note that in this scaling, the large- n limit is a regular function which is an eigenfunction of the harmonic oscillator (not a δ -function).